



SCSS-ST-SB-22  
PCM

Question Booklet Sr. No.

220042

Date : 10/04/2022

Time : 2.30 Hrs.

Marks : Section-A = 30 + Section-B = 240 = 270

**Important Instructions :**

1. Immediately fill the particulars on this page of the Test Booklet as well as Answersheet with Black or Blue Ball Pen. *Use of pencil is strictly prohibited.*
2. Do not open this Test Booklet until you are asked to do so.
3. This Test Booklet contains of 90 questions.
4. There are two sections in the question paper i.e. **Section-A** and **Section-B**.
5. The **Section-A** contains three parts i.e. Part-I, Part-II and Part-III.
6. The **Part-I** contains 10 questions of **English**.
7. The **Part-II** contains 10 questions of **Mental Ability**.
8. The **Part-III** contains 10 questions of **Basic Mathematics**.
9. In **Section-A**, each question carries **ONE** mark. There is no negative marking system.
10. The **Section-B** contains **THREE** parts i.e. Part-I, Part-II and Part-III.
11. The **Part-I** contains 20 questions of **Physics**.
12. The **Part-II** contains 20 questions of **Chemistry**.
13. The **Part-III** contains 20 questions of **Mathematics**.
14. In the **Section-B**, each question carries 4 marks. There is negative marking system. For each wrong answer 1 mark will be deducted.
15. There are four choices for every question, out of which only one choice is most correct.
16. Filling up more than one responses in any question will be treated as wrong response and marks for this will be deducted according to negative system.
17. No candidates is allowed to carry any printed or written textual material, bits of papers, cell phone and any other electronic devices.
18. Rough work is to be done on the space provided in the Test Booklet only.
19. On completion of the test, the candidate must hand over the Answer Sheet to the Invigilator on duty. *However, candidates are allowed to take away this Question Paper with them.*
20. **Do not fold or make any stray marks on the Answer Sheet.**

Name of the Candidate (in Capital letters) : \_\_\_\_\_

Seat No : In figures

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<b>Section - A : (Part-I – English)</b>
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**Instruction :**

1. Select the most suitable alternative which conveys the exact meaning in accordance with the correct grammatical rules.
- 1) My sister had left for America last week.
  - 2) My sister has been left for America last week.
  - 3) My sister has left for America last week.
  - 4) My sister left for America last week.

**Instruction :** Identify the part that contains an error.

2. Neither (A)/ the old man nor his (B) / children knows (C) / what to do about the (D) / problem.
- |      |      |
|------|------|
| 1) A | 2) B |
| 3) C | 4) D |

**Instruction :** Choose the one which best replaces the underlined words.

3. One of the function of a teacher is to spot cases of maladjustment.
- |                            |                            |
|----------------------------|----------------------------|
| 1) Most of the function of | 2) One of the functions of |
| 3) One of the functions by | 4) NO change               |

**Instruction :** Find the word that conveys the same meaning.

4. DOMAIN.
- |              |         |
|--------------|---------|
| 1) area      | 2) main |
| 3) marketing | 4) wild |

**Instruction :** Select the word that is opposite in meaning.

5. CONVEX.
- |            |            |
|------------|------------|
| 1) concave | 2) U-shape |
| 3) bent    | 4) arched  |

**Instruction :** Select the pair of words to replace the question mark.

6. Kindness : Mercy :: Cruelty : ?
- |               |                |
|---------------|----------------|
| 1) Savageness | 2) Wildness    |
| 3) Death      | 4) Destruction |

**Instruction :** Choose the one that can be substituted for the given phrase.

7. Teetotaler means.
- 1) one who abstains from theft
  - 2) one who abstains from meat
  - 3) one who abstains from taking wine
  - 4) one who abstains from taking malice

**Instruction :** Find out the correct meaning of the Idiomatic expression :

8. To be at the zenith of.
- 1) to die an immature death
  - 2) to be succumbed to one's flattery
  - 3) to be at the peak of
  - 4) face difficulty boldly

**Instruction :**

9. Choose the correctly spelt word :
- |               |                |
|---------------|----------------|
| 1) accomodate | 2) accommodate |
| 3) accomodate | 4) acomodate   |

**Instruction :** Choose the correct alternative out of the four choices to complete the sentence.

10. If I \_\_\_\_\_ you, I would have told him the truth.
- |        |         |
|--------|---------|
| 1) am  | 2) were |
| 3) was | 4) and  |



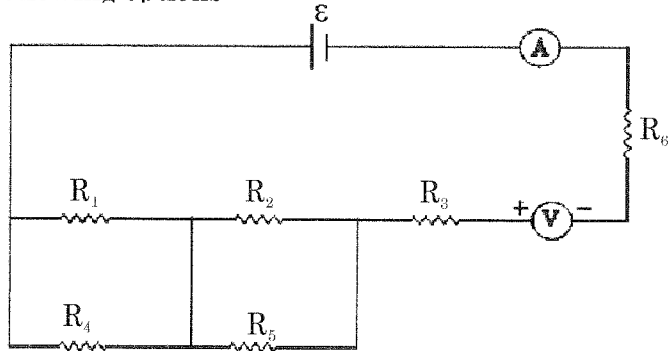




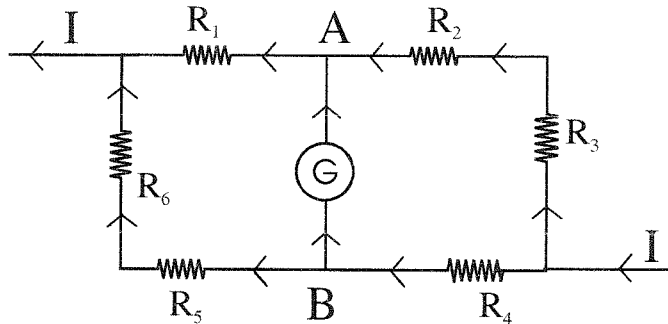




36. A coil of metal wire is kept stationary in a uniform magnetic field, then  
 1) an emf is induced in the coil      2) a current is induced in the coil  
 3) neither emf nor current is induced    4) both emf and current are induced
37. In the given circuit, the ammeter and the voltmeter are ideal. The reading of ammeter is almost zero, and voltmeter is not showing any reading, to have a finite current  $I$ , how do you proceed with the following options



- 1) Ammeter must be connected in parallel  
 2) Ammeter and voltmeter are to be interchanged  
 3) Voltmeter is not working  
 4) Voltmeter must be connected in parallel across any resistor
38. The galvanometer (G) in the circuit is a sensitive device and cannot sustain with a high current, to protect it, a high resistance (HR) is provided to you, choose the correct option

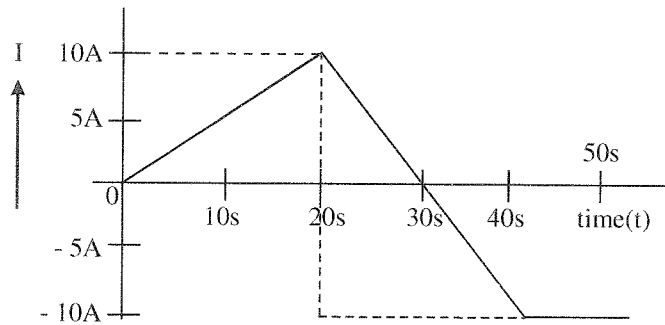


- 1) HR is not needed  
 2) HR must be connected in between A and G  
 3) HR must be connected in between G and B  
 4) HR can be connected anywhere in the branch AB

*Space For Rough Work*



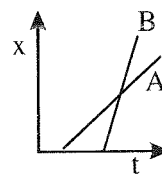
39. The current strength (I) in ampere on Y-axis and time (t) in second on X-axis plotted as shown in the figure. Analyse and choose the correct option of the following

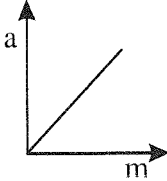
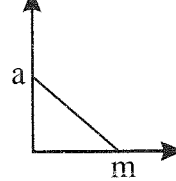
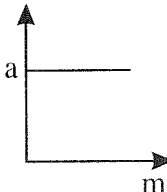
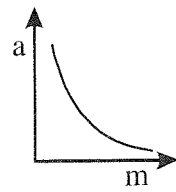


- 1) In the interval 0s to 20s current has same direction and magnitude
  - 2) The direction of the current is reversed just after 20 second
  - 3) The strength of the current is increasing from 30 second to 40 second, but in opposite direction to that of initial direction during 0 s to 30 s
  - 4) The directions of the current between 25 second to 30 second and 15 second to 20 second are opposite
40. In a liquid across a cross-section, the flow of charge (including positive and negative ions) within a time interval of one milli second is as shown, the current strength can be
- 1)  $4 \times 10^{-A}$  along +ve x-axis
  - 2) 10 A along +ve x-axis
  - 3)  $10^{-2}$  A along +ve x-axis
  - 4) 0.4 A along -ve x-axis
- 
41. For a particle at rest, which of the following quantity does not change any where
- 1) mass
  - 2) weight
  - 3) gravitational force
  - 4) acceleration due to gravity
42. Two objects of mass 2 kg and 8 kg separated by a distance of 4m. Find the gravitational force between the two bodies
- 1)  $4 \times 6.67 \times 10^{-11}$  N
  - 2)  $4 \times 6.67 \times 10^{-8}$  N
  - 3)  $6.67 \times 10^{-8}$  N
  - 4)  $6.67 \times 10^{-11}$  N
43. Find the increase in K.E. of a body of mass 200 g, when its speed increases from 4 m/s to 5 m/s ?
- 1) 0.9 J
  - 2) 9 J
  - 3) 900 J
  - 4) 90 J

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44. The kinematical equations of the motion can be applicable only when the particle has,
- 1) displacement constant
  - 2) variable acceleration
  - 3) acceleration constant
  - 4) non uniform acceleration
45. Prapti has applied a force of 50 N on an object, at an angle of  $30^\circ$  to the vertical. The object gets displaced in the horizontal direction and 300 J work is done. What is the displacement of the object ?
- 1) 12 m
  - 2)  $\frac{12}{\sqrt{3}}$  m
  - 3)  $\frac{12}{\sqrt{2}}$  m
  - 4) 6 m
46. A force is needed to
- i) change state of rest
  - ii) change state of motion
  - iii) change direction of motion
- Choose the correct option**
- 1) (i) only
  - 2) (i) and (ii) only
  - 3) (i)(ii) and (iii)
  - 4) (iii) only
47. Motion of the particle along the circumference of the circle is known as uniform circular motion, if its ----- remains constant
- 1) speed
  - 2) velocity
  - 3) acceleration
  - 4) displacement
48. Motion of the particle is known as non-uniform motion, if its
- I) speed constant
  - II) velocity constant
  - III) acceleration constant
  - IV) acceleration variable
- which of the following option is correct.**
- 1) only I
  - 2) only II
  - 3) only III
  - 4) III or IV
49. Figure shows distance time graphs of two objects A & B. Which object is moving with a greater speed when both are moving ?
- 1) A is faster than B
  - 2) B is faster than A
  - 3) Both A & B have same speed
  - 4) B is stationary



50. If a constant force is applied on bodies of different masses, corresponding acceleration – mass graph is
- 1) 
  - 2) 
  - 3) 
  - 4) 

<b>Part II – Chemistry</b>
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51. The water which is fit for drinking is called
- |                  |                 |
|------------------|-----------------|
| 1) black water   | 2) hot water    |
| 3) potable water | 4) sewage water |
52. The mass of an electron is about  $\frac{1}{X}$  times the mass of hydrogen atom. The value of X is
- |         |          |
|---------|----------|
| 1) 2    | 2) 200   |
| 3) 2000 | 4) 20000 |
53. What will be the percentage loss in mass when  $\text{NaHCO}_3$  is heated at  $30^\circ\text{C}$ ? (approximate)  
Molecular mass of  $\text{NaHCO}_3 = 84$ ,  $\text{Na}_2\text{CO}_3 = 106$
- |        |        |
|--------|--------|
| 1) 60% | 2) 45% |
| 3) 37% | 4) 70% |
54. Which of the following pair of oxides are amphoteric ?
- |                                      |   |
|--------------------------------------|---|
| 1) $\text{SO}_3, \text{CO}_2$        | 2) $\text{Al}_2\text{O}_3, \text{ZnO}$  |
| 3) $\text{Na}_2\text{O}, \text{MgO}$ | 4) $\text{Al}_2\text{O}_3, \text{CO}_2$ |
55. Which of the following molecule has maximum type of bonds ?
- |  |                     |
|--|---------------------|
| 1) $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ | 2) $\text{NaClO}_4$ |
| 3) HF  | 4) $\text{NH}_3$    |
56. Iron is harder than sodium because
- 1) iron atoms are smaller
  - 2) iron atoms are more closely packed
  - 3) metallic bonds are stronger in sodium
  - 4) metallic bonds are stronger in iron
57. At  $25^\circ\text{C}$  & 760 mm of Hg pressure a gas occupies 600 ml volume. What will be its pressure at a height where temperature is  $10^\circ\text{C}$  and volume of the gas is 640 ml ?
- |                |                 |
|----------------|-----------------|
| 1) 338.3 mm Hg | 2) 1353.2 mm Hg |
| 3) 676.6 mm Hg | 4) 1014.9 mm Hg |
58. Which of the following contains highest number of oxygen atoms.
- |                       |                               |
|-----------------------|-------------------------------|
| 1) 1g of Oxygen atom  | 2) 1g of $\text{O}_2$         |
| 3) 1g of $\text{O}_3$ | 4) 2g of $\text{H}_2\text{O}$ |
59. The acid used for washing eye is
- |                      |                |
|----------------------|----------------|
| 1) boric acid        | 2) acetic acid |
| 3) hydrochloric acid | 4) oxalic acid |
60. Which of the following has lowest melting point ?
- |  |                                   |
|--|-----------------------------------|
| 1) Acetic acid ( $\text{CH}_3\text{COOH}$ )      | 2) Chloroform ( $\text{CHCl}_3$ ) |
| 3) Ethanol ( $\text{CH}_3\text{CH}_2\text{OH}$ ) | 4) Methane ( $\text{CH}_4$ )      |
61. In  $\text{NaCl}$  ionic solid, each  $\text{Na}^+$  ion is surrounded by how many  $\text{Cl}^-$  ions?
- |      |      |
|------|------|
| 1) 6 | 2) 8 |
| 3) 4 | 4) 2 |
62. Among (a)  $\text{Na}_2\text{O}$  (b)  $\text{MgO}$  (c)  $\text{Al}_2\text{O}_3$  (d)  $\text{P}_2\text{O}_5$  (e)  $\text{Cl}_2\text{O}_7$  the most basic, most acidic and amphoteric oxide can be
- |            |            |
|------------|------------|
| 1) a, b, c | 2) b, e, c |
| 3) a, e, c | 4) e, c, a |

63. Lead nitrate on decomposition releases brown colour fumes. The chemical formula of brown colour fumes is
- 1)  $\text{NO}_2$
  - 2)  $\text{NO}$
  - 3)  $\text{N}_2\text{O}$
  - 4)  $\text{N}_2\text{O}_5$
64. When Bauxite is heated with  $\text{NaOH}$  solution. The water soluble compound formed is
- 1)  $\text{NaAlO}_2$
  - 2)  $\text{Na}_3\text{AlO}_3$
  - 3)  $\text{Al}(\text{OH})_3$
  - 4)  $\text{Al}_2\text{O}_3$
65. A student takes about 2 ml ethanoic acid in a dry test tube and adds a pinch of sodium carbonate to it. What will observe?
- 1) A colourless and odourless gas evolves with brisk effervescence.
  - 2) A colourless and odourless gas evolves which burns with pop sound when a burning candle is brought near it.
  - 3) A brown coloured gas with foul smell evolves with a brisk effervescence.
  - 4) A brown coloured gas with foul smell evolves which burns with pop sound when a burning candle is brought near it.
66. The property of direct bonding between atoms of the same element to form a chain is called
- 1) isomerism
  - 2) polymerization
  - 3) dehydration
  - 4) catenation
67. Members of a homologous series have similar
- 1) chemical properties
  - 2) physical properties
  - 3) molecular weight
  - 4) general molecular formula
68. Match the following :
- |                |  |
|----------------|--|
| A) Cooking gas | i) $\text{C}_{12}\text{H}_{22}\text{O}_{11}$         |
| B) Sugar       | ii) $\text{C}_{10}\text{H}_{16}\text{O}$             |
| C) Camphor     | iii) $\text{C}_6\text{H}_6$                          |
| D) Benzene     | iv) $\text{C}_3\text{H}_8 + \text{C}_4\text{H}_{10}$ |
- 1) A – iv, B – ii, C – iii, D – i
  - 2) A – iii, B – ii, C – i, D – iv
  - 3) A – iv, B – i, C – ii, D – iii
  - 4) A – ii, B – iii, C – iv, D – i
69. Higher homologue of methyl alcohol on complete oxidation gives
- 1)  $\text{CO}_2$  and  $\text{H}_2\text{O}$
  - 2)  $\text{CH}_3\text{CHO}$
  - 3)  $\text{HCOOH}$
  - 4)  $\text{CH}_3\text{COOH}$
70.  $\text{CH}_4 + \text{Cl}_2 \xrightarrow{\text{Sunlight}} \text{CH}_3\text{-Cl} + \text{H-Cl}$  is an example of
- 1) addition reaction
  - 2) substitution reaction
  - 3) combustion reaction
  - 4) reduction reaction

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### IMPORTANT DATES

#### ONLY For 10th Maharashtra State Board Students ( Except CBSE, ICSE And Other Board Students )

Screening Test 2022 : Pen Paper Based ( OFF LINE MODE EXAM ONLY )	Please Refer Website : <a href="http://www.junior-shahucollegelatur.org.in">www.junior-shahucollegelatur.org.in</a>
Registration for Screening Test - 2022	22 Feb. 2022 to 14 Mar. 2022
Last date for fee payment of registered Candidates	15 Mar. 2022 (13:00 IST)
Admit Card available for downloading ( Login your Account )	06 Apr. 2022 (17:00 IST)
<b>SCREENING TEST - 2022 (OFFLINE MODE ONLY)</b>	<b><u>10 April 2022</u></b>
Copy of candidate responses to be available on the website	14 Apr. 2022 (13:00 IST)
Online display of provisional answer keys	14 Apr. 2022 (13:00 IST)
Feedback and comments on provisional answer keys from the candidates	14 Apr. 2022 (13:00 IST) 15 Apr. 2022 (13:00 IST)
Online declaration of final answer keys	16 Apr. 2022 (11:00 IST)
Result of SCREENING TEST 2022	16 Apr. 2022 (15:00 IST) (On Candidate Individual Login)
Parent's Meet (PCM GROUP)	17 Apr. 2022 (11:00 IST) (For Selected and Wait List Candidate)
Parent's Meet (PCB GROUP)	17 Apr. 2022 (14:00 IST) (For Selected and Wait List Candidate)
Admissions : First List	18 Apr. 2022 (11:00 IST) 21 Apr. 2022 (17:00 IST)

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