



Shiv Chhatrapati Shikshan Sanstha, Latur

Shahu Screening Test 2026

SYLLABUS - PCB GROUP

परिशिष्ट 'अ'

Basic Mathematics And Quantitative Aptitude And Reasoning

1. Number system, HCF and LCM, Square and Square root, Cube and Cube root, Average, Ratios and Proportion, Partnership.
2. Algebraic Identities : Expansion of $(A+B)^2$, $(A-B)^2$, $(A+B)^3$, $(A-B)^3$, A^3-B^3 , A^3+B^3 & $(A+B+C)^2$
3. Discount and percentage : Profit & Loss (Percentage based)
4. Logarithms : Defination, Properties of logarithms.
5. Time & Work, Time, Speed and Distance, Problems on trains, Boats and Streams.
6. Arithmetic Progression and geomatric progression.
7. Mensuration Surface areas & volumes.
8. Trigonometric ratios of standard angles, its application.
9. Plane geometry : (Concept of line & angle, triangle, quadrilaterals and circle)
10. Co-ordinate geometry : Graph of line (co-ordinate system) , Collinear points.
11. Probability, Permutation and combination.
12. S.I.Units & symbols.
13. Basic knowledge of vector : Vector quantities, Direction of vector, type of vector.

REASONING

1. Syllogism reasoning.
2. Data sufficiency - Blood relation, Order and Ranking, Clock, Calender, Direction sence test, Distance, Coding Decoding, Seating arrangement.
3. Counting figure, Cube and Dice.
4. Venn diagram reasoning.

Physics

1. **Motion in One Dimension** : Motion of an Object, Displacement & Distance, Speed & Velocity, Uniform & Non-uniform Linear motion, Acceleration (+ ve, - ve & Zero acceleration), Effects of speed & direction on Velocity due to acceleration, Distance-Time graph (Uniform & Non-uniform motion), VelocityTime graph (Uniform Motion & Uniformly Accelerated Motion). Equation of motion using graphical method, Kinematical Equations , Free fall.
2. **Laws of Motion** : Contact & Non contact forces, Balanced & unbalanced forces , Newton's First, Second & Third law of motion & their applications, Momentum & Law of Conservation of Momentum. Pressure - atmospheric pressure (**NCERT 8th**)
3. **Work, Energy & Power** : Work done by constant force, Mechanical Energy (Kinetic energy & its expression, Potential energy & its expression & their Applications), Law of conservation of energy, Power.
4. **Gravitation**: Newton's universal law of gravitation, Acceleration due to gravity, Variation of acceleration due to gravity with height from surface of earth.
5. **Fluid Mechanics**: Pressure on solids & Liquids, Buoyancy, Archimedes Principle, Flotation.
6. **Sound**: Production of sound, Propagation of sound in medium, Frequency of sound waves, Velocity of sounds in different media, Reflection of sound, Applications of reflection of sounds (SONAR, Sound Board), Ultrasound & it's Application for detecting defects.
7. **Heat** : (**State Board 8th, 10th**) Heat & Temperature, Thermometer, Specific heat & calorimeter, effects of heat (Expansion of solids, Liquids & Gases- General Idea Only), Latent heat, Anomalous behavior of water, Heat Capacity & Specific heat capacity.
8. **Current Electricity** : Electric cells , EMF of Cell, Combination of cells, Potential & potential difference, Free electron & Electric current, Resistance & Ohm's law, Graphs of Current V/s Voltage, Resistivity, Conductors & Insulators, Electric symbols, Resistor in series & parallel, Electric Circuits, Electric Power.
9. **Effects of Electric Current**: Magnetic effect of electric current & their activities(straight conductor, circular loop, solenoid), Force on current carrying conductor due to Magnetic Field, Heating Effects of Electric current.
10. **Light** : Mirrors (Reflection): Laws of Reflection & Their Activities, Multiple reflections, Mirror & types of mirror, Spherical mirrors & images formed by them, Magnification due to spherical mirrors.
11. **Lens (Refraction)** : Refraction of light, Laws of refraction, Refractive index, Dispersion of light, Lenses, Ray diagram of refracted light, Sign convention.
12. Units and Physical quantities ,conversion of units

Chemistry

1. **Inside the atom :** Charged particle in matter , Types of substances, Dalton's atomic theory, Thomson's Plum pudding model of atom, Rutherford's nuclear model of atom and scattering experiment, Bohr's atomic model, Drawbacks of Dalton's atomic theory, Thomson's Plum pudding model, Rutherford's Nuclear model, Subatomic particles (electron, proton, neutron), Atomic number, mass number, isotopes and isobars, valency, Electronic configuration of elements, Nuclear reactor.
2. **Composition of Matter :** Characteristics of states of matter, evaporation, Types of elements, types of compounds, solution and concentration of solution, types of mixture, true and colloidal solution molecular formula and valency, dot and cross formula for writing chemical formula.
3. **Metals and Nonmetals :** Physical properties of metals and non metals , chemical properties of metals and non metals , Uses of metals and non metals
4. **Pollution :** Pollutants, Air Pollution, Green house effect, Acid rains, Water pollution, Prevention and control of pollution, soil pollution.
5. **Acids bases and salts :** Introduction, Indicator, Effects of acid and bases on litmus paper, properties of acids and bases and neutralization.
Arrhenius theory of acids and bases, Concept of Bronsted acid and base, Lewis acids and bases, classification of acids and bases, concentration of acids and bases, pH of solution, universal indicator, Reaction of acids and bases with metals metal oxides, Carbonates and bicarbonates, Types of Salts, Crystallisation of water, Electrical conductivity of ionic compounds Electrolysis of water.
6. **Chemical Change and Chemical bond :** Introduction, Natural and manmade chemical changes, Ionic bond, Covalent bond, Co-ordinate bond , valency and lone pair of electron.
7. **Chemical Reaction and equations :** Chemical reaction, Combination, Decomposition, Displacement, Double Displacement, Chemical Equation and balancing of chemical equation, Types of Chemical reactions, Exothermic and endothermic reactions, Factors affecting the rate of chemical reactions, Oxidation and reduction, Oxidation number, oxidising agent and reducing agent, Corrosion and Rancidity.
8. **Periodic Classification of elements :** Classification elements, Dobereiner's Triads, Newland's Law of Octaves, Mendeleev's Periodic table, Merits and demerits, Modern Periodic table and its structure, Groups, Periods and electronic configurations, Periodic trends in the modern periodic table, Valency, Atomic size, Ionic size, Metallic and Non metallic nature, Ionisation energy, Electronegativity.
9. **Study of Gas Laws :** Properties Of Gases, Liquids And Solids, Boyle'S Law, Charles' Law, Gas Equation, Absolute Zero Temperature, Standard Temperature, Pressure, N.T.P. And S.T.P. Dalton's law of partial pressure, Diffusion and Effusion of gases, Comparison study of ideal gas and real gas.
10. **Measurement of Matter (Mole Concept) :** Laws Of Chemical Combination, Law of conservation of mass, Atom - Shape, Mass, Valency, Molecular Mass, Atomic Mass, Formula Mass, Radicals, Ions, Mole Concept - Avogadro's Number, Calculation Of Moles, Mass, Atoms, No. Of Particles, Relation of mole with mass, number & volume.
11. **Carbon Compounds :** Valency, Catenation Of Carbon, Formation Of Double And Triple Bond, Isomerism Including Single, Homologous Series of Alkane, Alkene, Alkyne And Relation With Molecular Mass, Types of carbon and hydrogen (In terms of primary, secondary, tertiary and quaternary).
Nomenclature Of Simple Compounds Having Functional Groups Including Double Bond And Triple Bond, Alcohol, Aldehyde and acid, Bond line representation of organic compound.
Hydrocarbon, Method Of Preparation Of Alkane, Alkene And Alkyne And Chemical Properties And Uses Also, Combustion, addition, oxidation, substitution reactions.
Preparation Properties (Physical And Chemical Both) Of Alcohol (Ethanol) And Carboxylic Acid (Acetic Acid) Uses Of Alcohol And Acetic Acid. Soaps and detergents Method of cleaning action of soap.
Occurance of carbon, allotropes, Carbondioxide and Methane : Occurance, Properties and Uses.
12. **Synthetic Fibres and Plastics :** Natural and artificial fibres (Synthetic)
Types of synthetic fibres, Characteristics of synthetic fibres.
Plastics and its characteristics.
Plastic and environment. (Biodegradable, Non Biodegradable)

Biology

Unit I:- Diversity in living Organism

1. What is living ?

- What is the basis of classification.
- Need of classification..
- Binomial nomenclature, Rules of Binomial nomenclature. (Refer State Board Class X)
- The hierarchy of classification. (Refer State Board Class X)

2. Biological classification

- R.H. Whittaker system of classification.
- Protista..(Brief idea)
- Plantae In brief. Animalia in brief.
- Monera.(Brief idea)
- Fungi definition, nutrition, its major class.
- Virus,,viroid, lichen. (Brief idea)

3. Classification of plants.

- Basis of classification.
- General characteristic feature of following groups
- Thallophyta, Bryophyta, Pteridophyta.
- Gymnosperm, Angiosperm.

4. Animal classification (Refer State Board Class X).

- New method of system of animal classification.
- Different criteria for classification, Grade of organization, Body Organization, Germ layers, Body cavity, Body segmentation.
- From phylum porifera to chordata. Non chordates up to phylum level and chordates up to class level.

Unit:-II Cell biology and organisation

1. The cell

- What are living Organism?
- Structure of prokaryotic and, Eukaryotic cell in brief.
- What is Cell Made Up?
- Structure and difference between animal and plant Cell?
- Concept of cell theory. In brief.
- Plasma membrane or cell membrane and it's role
- Cell Wall.
- Nature of cytoplasm and cell organelles
- Endoplasmic reticulum. Its type and role.
- Golgi apparatus and it's function.
- Plastid, chloroplast, mitochondria.
- Lysosome , Vacuole
- Chromosomes structure in brief.
- Useful apparatus for studying cells. (Refer State Board Class VIII)

2. Cell cycle and cell division

- Cell cycle phases.
- Significance of mitosis.
- Significance of meiosis.
- Mitosis and its phases.
- Meiosis and its phases.

3. Biochemistry of living

- Bio molecule Concept
- Introductory concept of major biomolecules like carbohydrate, protein, fat.
- Metabolism

4. Morphology of angiospermic plant

- Structure and modification of root, stem and leaves in brief.
- structure of flower and their parts.
- Aestivation.
- Types of seed their structure.
- Placentation.
- Inflorescence and their major types.

5. Plant Tissue

- Meristematic tissue and its different type.
- Permanent tissue.
- Types of simple and complex tissue.

6. Animal tissue

- Types of epithelial tissue.
- Types of muscular tissue.
- Types of connectivity.
- Types of nervous tissue.

Unit:- III-I-Plant life processes.

1. Respiration.

- Aerobic and anaerobic respiration.
- Structure of ATP.
- Tri carboxylic acid cycle. (TCA)
- Energy from different food component.
- Living Organism and energy production.
- Glycolysis. (EMP)
- Electron transport chain. (ETC)
- Respiratory Quotient.(RQ)

2. Photosynthesis in higher plant.

- Definition redox reaction of photosynthesis.
- Photophosphorylation (light reaction) in brief.
- Significance of photosynthesis.
- Phases of mechanism of photosynthesis.
- Dark reaction. In brief.

3. Plant growth and development

- Concept of growth.
- growth and differentiation.
- Discovery and major role of hormones. (auxin,, gibberellin,cytokinin, abscisic acid and ethylene.).
- Difference between growth regulators and growth inhibitors
- Types of seed germination

Unit-III-II ANIMAL LIFE PROCESSES.

1. Nervous coordination.

- Structure of neuron.
- Central nervous system.
- Autonomous nervous system.
- Chemical control.
- Nervous system in humans.
- Peripheral nervous system..
- Reflex action..

2. Hormonal Co ordination

- Different endocrine gland in human.
- Thyroid gland and their hormone.
- Pituitary gland and their hormone.

3. Respiratory system in human.

- Respiratory system in human.
- External respiration.
- Cellular level respiration.
- Effect of Altitude on breathing.
- Structure of lungs.
- Internal respiration.
- Breathing.
- Asphyxiation, Emphysema, Hypoxia.

4. Digestion in human (Refer NCERT Class X).

- Digestive system in human.
- Physiology of digestion.
- Different organ of digestive system & their Histology.
- Glands related to the digestive system.
- Diseases related to the digestive system.

5. Excretion : Human Excretory system

6. Circulatory system

- Blood and lymph
- Blood vessels
- Structure and working of heart
- Circulation of Blood

Unit:- IV Reproduction BIOLOGY

1. Asexual reproduction in organism

- Binary fission, multiple fission, budding, fragmentation, regeneration, vegetative propagation, spore formation.
- Concept of grafting.
- Importance of asexual reproduction

2. Sexual reproduction In plant

- Typical flower and its part and their role.
- significance of pollination.
- Male and female gamete formation
- Pollination self and cross.
- Agents of pollination.
- Fertilization and double fertilization .

3. Sexual reproduction In human being.

- Male and female reproductive system.
- Menstrual cycle.
- Fertilization.
- Histology of testis and ovary..
- Gamete formation..
- Development and birth.

4. Reproductive health.

- Definition & concept of reproductive health major.
- Reproduction and modern technique. IVF.
- Sperm bank.

Unit:-V Genetics and general biology

1. Genetic basis of inheritance

- Concept of inheritance or heredity- Definition of heredity, hereditary traits in plant animal bacteria and human being.
- Important terms to understand the Mendelism – Dominant, recessive, homozygous, heterozygous, genotype, phenotype, genes and allelomorph.
- Mendelian procedure of crossing technique.
- Mendel's principle of heredity → monohybrid cross, and dihybrid cross with example and problem.
- Mendel's laws of inheritance → Law of dominance, law of segregation, law of independent assortment, backcross with respect to monohybrid and dihybrid, polygene, incomplete dominance, Co dominance, multiple allele.

2. Chromosomal basis of inheritance

- Chromosomal theory in brief, structure of chromosome, types of chromosome, HGP in brief.
- sex determination → In insect, fruit fly, human being and honey bee.
- Genetic disorder → Disorder due to chromosomal abnormalities, monogenic disorder, polygenic disorder, mitochondrial disorder.
- Pedigree.

3. Molecular basis of inheritance

- Structure of nucleotide.
- Structure of DNA.
- central dogma of protein synthesis.
- Translation in brief.
- Nature and bonds in polynucleotide.
- Replication of DNA.
- Structure of RNA and transcription.
- HGP in brief.

4. Introduction to biotechnology

- Biotechnology Concept
- Commercial application of biotechnology in the field of agriculture and medicine.
- Green, white and blue revolution.
- Change in agriculture management due to biotechnology
- Application of biotechnology in floriculture, nursery and forestry
- Agri- tourism.
- Main area and benefits in biotechnology.
- Tissue culture technique

5. Useful and harmful microbes

- Useful microorganism like Lactobacillus, Rhizobium, Yeast & antibiotic producing microorganism.
- Harmful microorganism like Clostridium and other microbes.
- Antibiotics and its classification (Refer State Board Class IX)

6. Health and disease.

- Health immunity.
- Disease and its causes.
- Infectious and non infectious diseases.
- Means of spreading of disease.
- Government Scheme for vaccin and vaccination. (Refer State Board Class X)
- Types of immunity.
- Acute and chronic diseases.
- Disease causing agents.
- Treatment and prevention (T.B, typhoid hepatitis, rabies, polio. AIDS, diarrhoea.)

7. Evolution

- Different theory of evolution
- Evidences of evolution. Morphological anatomical paleontological embryological vestigial organ
- Homologous and analogous organ.
- Darwin's theory of natural selection.
- Speciation.
- Tracing evolutionary relationship.
- Connecting link.
- Lamarckism.
- Human evolution in brief.
- Fossils.

8. Energy flow in an ecosystem.

- Food chain and food web.
- Energy flow and it's importance.
- Energy pyramid.
- Producers, Consumers and decomposers.



Shiv Chhatrapati Shikshan Sanstha, Latur

Shahu Screening Test 2026

SYLLABUS - PCM GROUP

परिशिष्ट 'ब'

Mathematics

1. **Number Theory :** Real Numbers, Prime numbers, co-prime, relatively prime numbers, H.C.F. & L.C.M
2. **Linear Equations and inequation :** One variable linear equation, Two variable linear equation and geometrical interpretation, linear inequation and its application.
3. **Polynomials :** Zeros and roots of polynomial, geometrical meaning of roots, Remainder theorem, Factor theorem, Division of polynomial, Factorisation
4. **Expansion formulae :** Algebraic identities $(x \pm y)^2 = x^2 \pm 2xy + y^2$
 $(x+y+z)^2 = x^2 + y^2 + z^2 + 2xy + 2yz + 2zx$ $(x \pm y)^3 = x^3 \pm y^3 \pm 3xy(x \pm y)$
 $x^3 \pm y^3 = (x \pm y)(x^2 \mp xy + y^2)$ $x^3 + y^3 + z^3 - 3xyz = (x + y + z)(x^2 + y^2 + z^2 - xy - yz - zx)$
5. **Theory of equation and nth degree equation :** Quadratic Equations - Nature of Roots, Solution of Equations, Solution by factorisation, Relation between roots & coefficients of the equation, Cubic equations and its roots.
6. **Rationalization and Simplification** of the type $\frac{1}{a+b\sqrt{x}}$ and $\frac{1}{\sqrt{x}+\sqrt{y}}$, $\sqrt{x + \sqrt{ax - b}}, \sqrt{x - \sqrt{ax - b}}$
7. **Exponents & Powers :** Square & Square root, Cube & cube roots, Indices & Surds
 $a^m \times a^n = a^{m+n}$, $a^m \div a^n = a^{m-n}$, $(a^m)^n = a^{mn}$ etc.
8. **Ratio & Proportion :** Properties of ratio, Proportionality. Componendo, dividendo, alternendo, invertendo, properties and their combination.
9. **Sequence and series :** Progressions, nth term of A.P. , Sum of nth term of an A.P., Geometric progression (G.P.) nth term of G.P. & sum of nth term of G.P., Harmonic progression & its nth term, rules of sum of summation.
10. **Plane Geometry :** Lines and Angles, Polygons, Triangles, congruency of triangles (SSS, SAS, AAS and RHS), similarity of triangles (SSS, SAS, AA). Mid-point theorem and its converse, equal intercept theorem, Pythagoras theorem. Apollonius theorem.
 - Quadrilaterals - Its types, cyclic quadrilaterals and its properties,
 - Parallel lines - Perpendicular lines, Intersecting lines & properties.
 - Circles - Definition of circle in terms of locus. Its angle/arc properties, chord properties, cyclic properties, tangent, normal & secant properties.
11. **Trigonometry :** Trigonometric ratios, Values of trigonometric ratio, Trigonometric identities, Complimentary and Supplimentary angles, Heights & Distances.
12. **Mensuration :** Area & perimeter of triangle (Heron's formula), area and circumference of circle, area of sector of circle.
 Area (Surface area & total surface and curved surface area) Volume of 3D solids and figures - Cube, Cuboids, right circular Cylinder, right circular Cone, Sphere.
13. **Coordinate Geometry :** Distance Formula, Section formula, Area of triangle & collinearity , - Slope/ Gradient / $\tan \theta$ of line.
 - Equation of line. Slope - Intercept form $y = mx + c$. Two point form $(y - y_1) = m(x - x_1)$.
 - Geometric understanding of x-intercept & y-intercept and gradients.
 - Conditions for two lines to be parallel or perpendicular.
14. **Probability :** Random experiments, sample space, events, Classical definition of probability, Problems on dice, coin and playing card etc.
15. **Statistics :** Types of data set , random, grouped and ungrouped data.
 - Measure of central tendency, Mean, Mode, Median, Empirical relation between mean, median, mode
16. **Sets :** Type of sets, intervals, inequality.
17. **Logarithmic and exponential function :** Definition, relation between logarithm and exponential function. Properties of logarithm.
18. **Graph of Function :** Modulus function, quadratic expression, logarithmic function and exponential function.

Basic Mathematics And Quantitative Aptitude And Reasoning

1. Number system, HCF and LCM, Square and Square root, Cube and Cube root, Average, Ratios and Proportion, Partnership.
2. Algebraic Identities : Expansion of $(A+B)^2$, $(A-B)^2$, $(A+B)^3$, $(A-B)^3$, A^3-B^3 , A^3+B^3 & $(A+B+C)^2$
3. Discount and percentage : Profit & Loss (Percentage based)
4. Logarithms : Defination, Properties of logarithms.
5. Time & Work, Time, Speed and Distance, Problems on trains, Boats and Streams.
6. Arithmetic Progression and geomatric progression.
7. Mensuration Surface areas & volumes.
8. Trigonometric ratios of standard angles, its application.
9. Plane geometry : (Concept of line & angle, triangle, quadrilaterals and circle)
10. Co-ordinate geometry : Graph of line (co-ordinate system) , Collinear points.
11. Probability, Permutation and combination.
12. S.I.Units & symbols.
13. Basic knowledge of vector : Vector quantities, Direction of vector, type of vector.

REASONING

1. Syllogism reasoning.
2. Data sufficiency - Blood relation, Order and Ranking, Clock, Calender, Direction sence test, Distance, Coding Decoding, Seating arrangement.
3. Counting figure, Cube and Dice.
4. Venn diagram reasoning.

Physics

1. **Motion in One Dimension :** Motion of an Object, Displacement & Distance, Speed & Velocity, Uniform & Non-uniform Linear motion, Acceleration (+ ve, - ve & Zero acceleration), Effects of speed & direction on Velocity due to acceleration, Distance-Time graph (Uniform & Non-uniform motion), VelocityTime graph (Uniform Motion & Uniformly Accelerated Motion). Equation of motion using graphical method, Kinematical Equations , Free fall.
2. **Laws of Motion :** Contact & Non contact forces, Balanced & unbalanced forces , Newton's First, Second & Third law of motion & their applications, Momentum & Law of Conservation of Momentum. Pressure - atmospheric pressure (**NCERT 8th**)
3. **Work, Energy & Power :** Work done by constant force, Mechanical Energy (Kinetic energy & its expression, Potential energy & its expression & their Applications), Law of conservation of energy, Power.
4. **Gravitation:** Newton's universal law of gravitation, Acceleration due to gravity, Variation of acceleration due to gravity with height from surface of earth.
5. **Fluid Mechanics:** Pressure on solids & Liquids, Buoyancy, Archimedes Principle, Flotation.
6. **Sound:** Production of sound, Propagation of sound in medium, Frequency of sound waves, Velocity of sounds in different media, Reflection of sound, Applications of reflection of sounds (SONAR, Sound Board), Ultrasound & it's Application for detecting defects.
7. **Heat : (State Board 8th, 10th)** Heat & Temperature, Thermometer, Specific heat & calorimeter, effects of heat (Expansion of solids, Liquids & Gases- General Idea Only), Latent heat, Anomalous behavior of water, Heat Capacity & Specific heat capacity.
8. **Current Electricity :** Electric cells, EMF of Cell, Combination of cells, Potential & potential difference, Free electron & Electric current, Resistance & Ohm's law, Graphs of Current V/s Voltage, Resistivity, Conductors & Insulators, Electric symbols, Resistor in series & parallel, Electric Circuits, Electric Power.
9. **Effects of Electric Current:** Magnetic effect of electric current & their activities(straight conductor, circular loop, solenoid), Force on current carrying conductor due to Magnetic Field, Heating Effects of Electric current.
10. **Light :** Mirrors (Reflection): Laws of Reflection & Their Activities, Multiple reflections, Mirror & types of mirror, Spherical mirrors & images formed by them, Magnification due to spherical mirrors.
11. **Lens (Refraction) :** Refraction of light, Laws of refraction, Refractive index, Dispersion of light, Lenses, Ray diagram of refracted light, Sign convention.
12. Units and Physical quantities , conversion of units

Chemistry

1. **Inside the atom :** Charged particle in matter , Types of substances, Dalton's atomic theory, Thomson's Plum pudding model of atom, Rutherford's nuclear model of atom and scattering experiment, Bohr's atomic model, Drawbacks of Dalton's atomic theory, Thomson's Plum pudding model, Rutherford's Nuclear model, Subatomic particles (electron, proton, neutron), Atomic number, mass number, isotopes and isobars, valency, Electronic configuration of elements, Nuclear reactor.
2. **Composition of Matter :** Characteristics of states of matter, evaporation, Types of elements, types of compounds, solution and concentration of solution, types of mixture, true and colloidal solution molecular formula and valency, dot and cross formula for writing chemical formula.
3. **Metals and Nonmetals :** Physical properties of metals and non metals , chemical properties of metals and non metals , Uses of metals and non metals
4. **Pollution :** Pollutants, Air Pollution, Green house effect, Acid rains, Water pollution, Prevention and control of pollution, soil pollution.
5. **Acids bases and salts :** Introduction, Indicator, Effects of acid and bases on litmus paper, properties of acids and bases and neutralization.
Arrhenius theory of acids and bases, Concept of Bronsted acid and base, Lewis acids and bases, classification of acids and bases, concentration of acids and bases, pH of solution, universal indicator, Reaction of acids and bases with metals metal oxides, Carbonates and bicarbonates, Types of Salts, Crystallisation of water, Electrical conductivity of ionic compounds Electrolysis of water.
6. **Chemical Change and Chemical bond :** Introduction, Natural and manmade chemical changes, Ionic bond, Covalent bond, Co-ordinate bond , valency and lone pair of electron.
7. **Chemical Reaction and equations :** Chemical reaction, Combination, Decomposition, Displacement, Double Displacement, Chemical Equation and balancing of chemical equation, Types of Chemical reactions, Exothermic and endothermic reactions, Factors affecting the rate of chemical reactions, Oxidation and reduction, Oxidation number, oxidising agent and reducing agent, Corrosion and Rancidity.
8. **Periodic Classification of elements :** Classification elements, Dobereiner's Triads, Newland's Law of Octaves, Mendeleev's Periodic table, Merits and demerits, Modern Periodic table and its structure, Groups, Periods and electronic configurations, Periodic trends in the modern periodic table, Valency, Atomic size, Ionic size, Metallic and Non metallic nature, Ionisation energy, Electronegativity.
9. **Study of Gas Laws :** Properties Of Gases, Liquids And Solids, Boyle's Law, Charles' Law, Gas Equation, Absolute Zero Temperature, Standard Temperature, Pressure, N.T.P. And S.T.P. Dalton's law of partial pressure, Diffusion and Effusion of gases, Comparison study of ideal gas and real gas.
10. **Measurement of Matter (Mole Concept) :** Laws Of Chemical Combination, Law of conservation of mass, Atom - Shape, Mass, Valency, Molecular Mass, Atomic Mass, Formula Mass, Radicals, Ions, Mole Concept - Avogadro's Number, Calculation Of Moles, Mass, Atoms, No. Of Particles, Relation of mole with mass, number & volume.
11. **Carbon Compounds :** Valency, Catenation Of Carbon, Formation Of Double And Triple Bond, Isomerism Including Single, Homologous Series of Alkane, Alkene, Alkyne And Relation With Molecular Mass, Types of carbon and hydrogen (In terms of primary, secondary, tertiary and quaternary).
Nomenclature Of Simple Compounds Having Functional Groups Including Double Bond And Triple Bond, Alcohol, Aldehyde and acid, Bond line representation of organic compound.
Hydrocarbon, Method Of Preparation Of Alkane, Alkene And Alkyne And Chemical Properties And Uses Also, Combustion, addition, oxidation, substitution reactions.
Preparation Properties (Physical And Chemical Both) Of Alcohol (Ethanol) And Carboxylic Acid (Acetic Acid) Uses Of Alcohol And Acetic Acid. Soaps and detergents Method of cleaning action of soap.
Occurance of carbon, allotropes, Carbondioxide and Methane : Occurance, Properties and Uses.
12. **Synthetic Fibres and Plastics :** Natural and artificial fibres (Synthetic)
Types of synthetic fibres, Characteristics of synthetic fibres.
Plastics and its characteristics.
Plastic and environment. (Biodegradable, Non Biodegradable)